

# NMR Short Course 2010

## Problem Set I

Read:

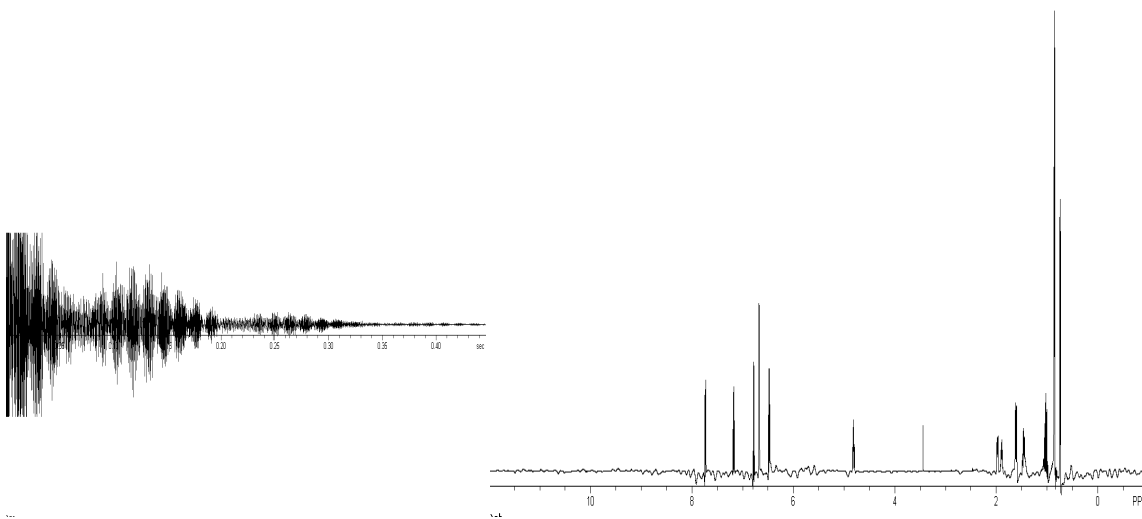
Chapter 2 --- *High Resolution NMR Techniques in Organic Chemistry*

1. Natural carbon ( $Z = 6$ ) is composed of two isotopes,  $^{12}\text{C}$  and  $^{13}\text{C}$ . The atomic mass listed for carbon in the periodic table is 12.01. (1) What is the composition of each nucleus? (2) What is the natural abundance of each isotope? Note: General question for Chemistry
  
2. What is the energy difference between the two spin states of  $^1\text{H}$  in magnetic field of 9.395 Tesla (400MHz instrument) and 18.79 Tesla (800MHz instrument), How about  $^{13}\text{C}$ ? ( $\gamma_{\text{H}} = 267.512 \times 10^6 \text{ rad T}^{-1} \text{ S}^{-1}$  and  $\gamma_{\text{C}} = 67.2640 \times 10^6 \text{ rad T}^{-1} \text{ S}^{-1}$ ). How much intensity will increase by using 800 MHz instrument, instead of using 400 MHz instrument. Note: Read book page 12.

$$\Delta E = \frac{\gamma h B}{2\pi}$$

3. (1). At 20  $^{\circ}\text{C}$  what fraction of  $^1\text{H}$  nuclei in 9.395 T field are the upper and lower states? (2). of  $^{13}\text{C}$ ? Note: read book page 12-13.

4. (a). Suppose that for a certain set of  $^{13}\text{C}$  nuclei at  $25^\circ\text{C}$ , the value of  $T_1$  is 2 s. How long after immersion in a 9.395 Tesla magnetic field will it take for an initially equal distribution of  $^{13}\text{C}$  spin states to progress 95% of the way toward equilibrium? (b). What would happen if the magnet were turned off at this point? Note: Read "High-Resolution NMR Techniques in Organic Chemistry" page 21 section 2.4.1
5. Explain what effect dissolved Oxygen ( $\text{O}_2$ ) might have on longitudinal relaxation of  $^1\text{H}$  nuclei. *Hint: the oxygen molecule has two unpaired electrons with the same  $s$  value.* Note: General chemistry Question.
6. The following spectrum was acquired on our INOVA400. Explain what is happened and to change which parameters will make the spectrum better? Note: Read book. Page 51.



6. Following two  $^1\text{H}$  spectra of strychnine acquired on INOVA400 and INOVA600. (1). Calculate the chemical shifts of four labeled signals from both instruments. (2) Calculate the homonuclear J coupling of two protons using the data on the spectra.

